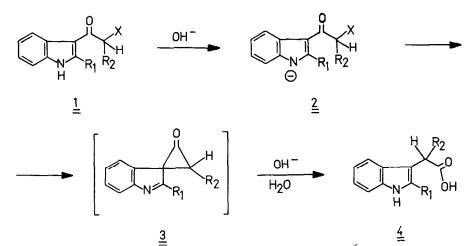
BASE-INDUCED REACTIONS OF 3-(a-HALOACYL)INDOLES Jan Bergman and Jan-Erling Bäckvall Department of Organic Chemistry, Royal Institute of Technology, S-100 44 Stockholm 70, Sweden

(Received in UK 25 May 1973; accepted for publication 20 June 1973)

Forcing conditions, such as powdered sodium hydroxide in refluxing xylene^{1,2} or silver nitrate in ethanol,³ are necessary to effect Favorskii rearrangements of aryl α -haloalkyl ketones (RCHXCOAr \longrightarrow RCHArCOOH). With 3-(α -haloacyl)indoles, which should be relatively strong acids,⁴ as reactants we anticipated the formation of indoleacetic acids under mild conditions along the following reaction path:



The reactions were carried out as follows: $3-(\alpha-haloacyl)indole^{6}$ (0.02 mol) was added in ten portions to a boiling solution of NaOH (2.4 g 0.06 mol) in 80 % ethanol (200 ml) during l h. After another hour of reflux the acid was extracted in moderate yield. Competitive substitutions (RCHXCOAr \longrightarrow RCHOHCOAr + RCHOEtCOAr) explain the low yields (<u>cf</u>. ref. 7). The yields of substitution products varied in the range of 30 - 80 %.

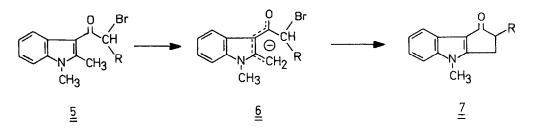
		Yield ^{a} (%)	M.p. ^o C	lit. m.p.
<u>4a</u>	R ₁ =H, R ₂ =CH ₃	15, 11 ^b	105-7	102 ^d , 113-114 ^e
<u>4</u> 2	^R 1 ^{=H} , ^R 2 ^{=C} 2 ^H 5	12	106-7	106 ^e
<u>4</u> ⊆	^R 1 ^{=H} , ^R 2 ^{=C} 6 ^H 5	18 ⁰	184-86 ^f	
<u>4</u> ₫	R ₁ =CH ₃ , R ₂ =CH ₃	20	135-37 ⁹	

Yields of indole-3-acetic acids $(\underline{4})$

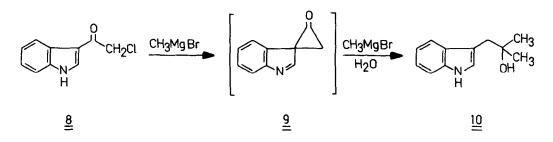
^a \downarrow (X=Br) was used as starting material unless stated otherwise. ^b \downarrow (R₁=H, R₂=CH₃, X=C1) was used as starting material. ^c \downarrow (R₁=H, R₂=C₆H₅, X=C1) was used as starting material. ^d F. Kögl and D. Kostermans, Z. *Physiol. Chem.*, <u>235</u>, 201 (1935). ^e K.K. Schlender, M.J.Bukovac, and H.M. Sell, *Phytochemistry*, <u>5</u>, 133 (1966). ^f Calc. for: C₁₆H₁₃NO₂: C, 76.5; H, 5.2; N, 5.6. Found: C, 76.4; H, 5.2; N, 5.7. ^g Calc. for: C₁₂H₁₃NO₂: C, 71.0; H, 6.5; N, 6.9. Found: C, 71.1; H, 6.6; N, 7.1.

Although the yields of indoleacetic acids are low to moderate, we consider this rapid procedure suitable for the preparation of these compounds for laboratory purposes (<u>e.g.</u> for biological studies). The starting materials ($\underline{1}$) are readily available⁶ through acylation of indoles by a pyridine α -haloacyl halide reagent.

In accordance with the suggested mechanism <u>N</u>-methylated 3-(α -haloacyl)indoles could not be transformed to Favorskii rearrangement products. The attempted rearrangement of $\frac{5}{2}$ (R=H) and $\frac{5}{2}$ (R=CH₃) gave the cyclization products $\underline{7}$ (R=H) and $\underline{7}$ (R=CH₃) respectively, presumably <u>via</u> the resonance-stabilized anion $\underline{6}$. Compound $\underline{7}$ (R=CH₃) was formed in good yield whereas the yield of $\underline{7}$ (R=H) was low due to competitive substitution. The structure of $\underline{7}$ (R=H) was proved by transformation (LAH reduction) to the known⁸ compound <u>N</u>-methyl-2,3-cyclopentanoindole.



The spirocyclic cyclopropane $(\underline{3})$ should be an expected intermediate in several reactions performed under basic conditions. In accordance with this 3-chloroacetylindole $(\underline{8})$ when treated with $CH_{3}MgI$ gave α, α -dimethyltryptophol $(\underline{10})$ identical with an authentic sample.^{9,10} Interestingly enough Bohlmann and Kapteyn¹¹ have similarly transformed 4-hydroxy- ω -chloracetophenone to 4-hydroxybenzyldimethylcarbinol.

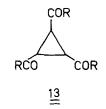


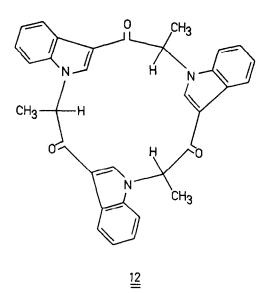
LAH-Reduction of $\underline{\delta}$ similarly gave tryptophol (30 % yield) together with 3-ethylindol ($\underline{\underline{1}}$). The latter compound is probably formed as follows ($\underline{\delta}$ ------ 3-Acetylindole ------ $\underline{\underline{1}}$).¹²

When sodium hydroxide (1.5 g) in 80 % ethanol (60 ml) was added to a boiling solution of 3-(α -brompropionyl)-indole (2.52 g, 0.01 mol) during 1 h (<u>e.g.</u> reversed addition), it resulted in a complex (polymeric) mixture. From this mixture one defined product could be isolated (yield: 12 %). From the mass spectroscopic data given below we have assigned it structure <u>l</u>2. The absence of NH vibrations and the presence of strong C=O vibrations at 1658 cm⁻¹ further supported structure <u>l</u>2.

MS of <u>12</u>, <u>m/e</u> (rel. intensity) 513 (36, M); 498 (10); 342 (27); 341 (100); 171 (29); 170 (60); 144 (18); 143 (25); 128 (10); 78 (13).

Interestingly Sanna¹⁴ has isolated a trimeric product from the reaction of 2-methyl-3-chloroacetylindole with ammonia at 100° . The Italian worker gave structure <u>13</u> (R = 2-methylindol-3-yl) to this product. We will reinvestigate this reaction in connection with further studies of <u>12</u>.





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